

molekō™

TESSENDERLO GROUP

CYNTROL®

REFINERY

CYANIDE

CONTROL

PROGRAM

MOLECULES ON THE MOVE



CYNTROL®

(AMMONIUM

POLYSULFIDE

SOLUTION)

Works in refinery applications to convert cyanide into a non-corrosive thiocyanate, protecting equipment from corrosion.

Catalytic cracking and coking operations in a refinery typically generate some amount of cyanide due to the presence of nitrogen containing compounds in crude oil. These cyanides, if present in sufficient concentration, can have a deleterious effect on refinery operations, due to their promotion of **hydrogen related corrosion**.

While other corrosion inhibitors, such as filmers, protect equipment by forming a barrier between the metal and the corrosive agent, the cyanide still exists and is passed along to other units within the refinery. Cyntrol effectively eliminates the cyanide by converting it to a **non-corrosive thiocyanate** that is easily handled in the refinery's water treatment facility.

BENEFITS

- Prevents cyanide from attacking the passivated iron sulfide layer in refinery equipment
- Strengthens protective passivated layer by converting FeS to FeS₂
- Extends life of expensive refinery equipment
- Improves efficiency of wash water system, as more cyanide can be solubilized into the wash water
- Untreated cyanide can lead to reduced efficiency in the amine system through formation of heat stable salts
- Reduces potential environmental issues

TYPICAL PROPERTIES

	Cyntrol 2040	Cyntrol 2045
Active Ingredient (ammonium polysulfide)	49%	55%
Specific Gravity	1.125 – 1.150	1.145 – 1.175
Appearance	Ruby red liquid	Ruby red liquid
pH	10.8 – 11.5	10.8 - 11.5
Vapor Pressure	41.86 kPa @ 68°F (314 mm Hg at 20°C)	41.86 kPa @ 68°F (314 mm Hg at 20°C)
Freezing Point	5°F (-15°C)	5°F (-15°C)
Boiling Point	100°F (38°C)	100°F (38°C)

SOUR WASH WATER

PRE-TREATMENT

Iron sulfide (FeS) visible from cyanide corrosion



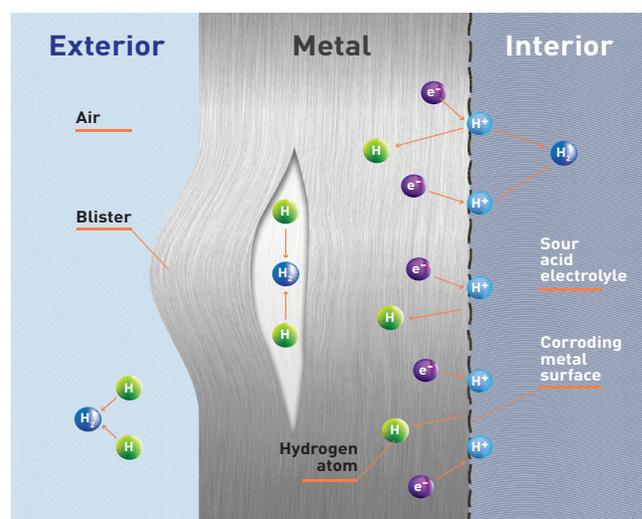
POST-TREATMENT

FeS-free, Cyntral treated

HYDROGEN BLISTERING CORROSION

In addition to removing the passivated iron sulfide layer, the cyanide, along with the hydrogen sulfide (H_2S) also present in refinery sour wash water, suppress the formation of molecular hydrogen (H_2) thereby allowing the atomic hydrogen to diffuse readily into a metal's structure. This process diminishes the mechanical integrity of many metals and alloys and leads to hydrogen induced cracking (HIC).

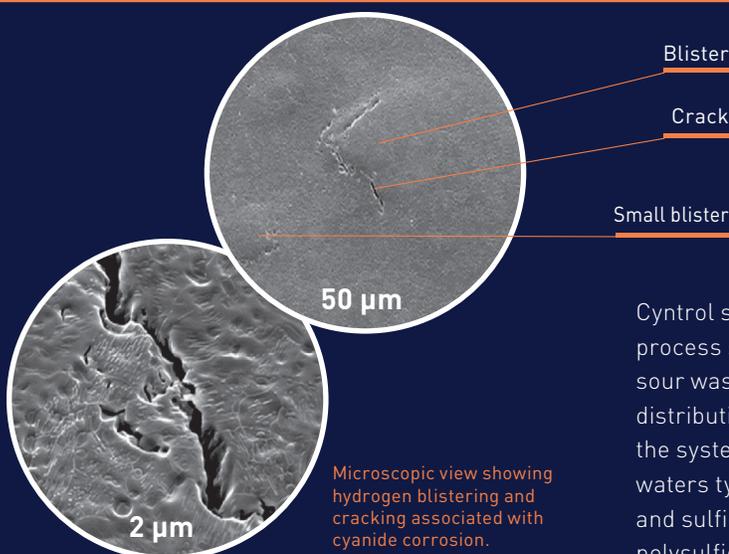
HIC is generally first evident as blister formation in the vessel inner surface. Cracking can then occur when the atomic hydrogen precipitates as H_2 and the resulting pressure exceeds the tensile strength of the already diminished metal structure.



Hydrogen blistering in the wall of a container.

<https://faculty.kfupm.edu.sa/me/hussaini/corrosion%20engineering/Images/Hydrogen-wallblister.jpg>

Cyntral is typically suitable for addition as part of system wash waters, at moderate temperatures (50 – 230°F) in a slightly basic environment (pH of 8.0 – 10.0). The reaction between the polysulfide and cyanide ions is an aqueous phase reaction.



Microscopic view showing hydrogen blistering and cracking associated with cyanide corrosion.

Cyntral should be injected into process streams containing sour wash water to provide good distribution of Cyntral through the systems to be treated. Sour waters typically contain ammonia and sulfides which help keep polysulfides in solution.

DOSAGE

Dosage is typically adjusted to convert all cyanide to thiocyanate while maintaining a small residual of Cyntrol (roughly 10 PPM) in the wash water.

A general starting dosage guideline is about 0.00135 gallons per hour of Cyntrol is needed per GPM of wash water containing one PPM of cyanide.

$$\begin{array}{|c|} \hline \text{Cyntrol} \\ \text{Dosage} \\ \text{(gal/hr)} \\ \hline \end{array} = \begin{array}{|c|} \hline \text{Wash} \\ \text{water} \\ \text{(GPM)} \\ \hline \end{array} \times \begin{array}{|c|} \hline \text{Cyanide} \\ \text{level} \\ \text{(PPM)} \\ \hline \end{array} \times \begin{array}{|c|} \hline 0.00135 \\ \hline \end{array}$$

For example, a 40 GPM flow with a 25 PPM cyanide level would require $40 \times 25 \times 0.00135 = 1.35$ gallons/hr Cyntrol injection.

PACKAGING AND HANDLING

- Product is available in bulk tank trucks for transfer to on-site storage tanks
- Storage tank considerations include:
 - An appropriate nitrogen pad, or system venting through an appropriate scrubber system
 - A barrier fluid to minimize vapor generation
 - Avoid direct sunlight and incompatible materials/chemicals
- Maintain at moderate temperatures in a well-ventilated area
- Avoid contact with skin and eyes
- Wash thoroughly after handling
- Observe all safety precautions listed in the safety data sheet (SDS).

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SURVEY SERVICES

moleko offers on-site technical survey services to help determine if refinery processing units are producing excess cyanide. By testing sour water samples from various points throughout the refinery, technical staff can make informed decisions about where cyanides are generated and where they are being moved, impact of cyanide on the refinery, effectiveness and efficiency of wash water system, and make recommendations for process improvements. A summary report of the results and observations is provided following the survey. Typical results include WAD cyanide, pH, ammonia, sulfides, thiocyanates (if currently using Cyntrol), and physical appearance (color, turbidity, solids, and hydrocarbons).

Cyntrol significantly reduces the potential for cyanide assisted corrosion and the presence of cyanide in the refinery's waste water effluent.

CONTACT US FOR A TECHNICAL ASSESSMENT



SCAN ME